

Contents lists available at ScienceDirect

Journal of Fluorine Chemistry



journal homepage: www.elsevier.com/locate/fluor

Towards functional fluorous surfactants. Synthesis of hydrophilic fluorous 1,2,3-triazolylmethyl ethers and di(1,2,3-triazolylmethyl) ethers

Dominic V. Francis, D. Howard Miles¹, Adnan I. Mohammed², Roger W. Read^{*}, Xiaobei Wang

School of Chemistry, University of New South Wales, UNSW, Sydney, NSW 2052, Australia

ARTICLE INFO

Article history: Received 1 April 2011 Received in revised form 30 June 2011 Accepted 4 July 2011 Available online 12 July 2011

Keywords: Fluorous chemistry Microwave acceleration Dipolar cycloaddition Fluorous 1,2,3-triazoles Triazolylmethyl ethers Fluorous surfactants

ABSTRACT

Copper(I)-accelerated Huisgen–Meldal dipolar cycloaddition reactions between polyfluoroalkyl azides and propargyl ethers of *n*-octanol and of triethyleneglycol monomethyl ether exhibited variation in yield of 1,2,3-triazol-4-ylmethyl ethers. Microwave acceleration, and *in situ* generation of the azides, provided improvements in yield and efficiency. In contrast, very good yields of equivalent fluorous triazoles were obtained from a range of *n*-alkyl azides with propargyl ethers of perfluorohexylethanol and of perfluoroheptylmethanol through conventional copper(I)-promoted reactions. Together, the resulting substances with systematic variations in polyfluoroalkyl and alkyl substituent length and position of substitution, and degree of oxygen content, make up small libraries of hybrid fluorous 1,2,3-triazol-4-ylmethyl ethers as candidates for study as hydrophilic fluorous surfactants. In addition, a pilot sample of di(1,2,3-triazol-4-ylmethyl) ethers with 1'-octyl-1-polyfluoroalkyl-substituents and 1'-nonyl-1-per-fluorooctylethyl substituents were synthesised for the first time in an effort to develop more functional, fluorous surfactants.

© 2011 Elsevier B.V. All rights reserved.

1. Introduction

There has been widespread interest in polymeric and other, hybrid fluorous compounds because of the unique surface activity and physical and biological properties of materials derived from them [1-10]. Implicit in many reports has also been the potential of the compounds for the carriage of added functionality [11]. Fluorous modifications of small molecules derived from heterocycles have received less systematic attention as surfactants [11-19], although many innovative applications of fluorous tagged molecules have appeared [20], especially in terms of separation strategies [21] and recyclable catalysts [22,23]. We have recently outlined the potential of small fluorous heterocycles as surfactants and participants in self-assembly [24], and have described the synthesis of 1.2.3-triazoles [24.25] and tetrazoles [25] with one perfluoroalkyl substituent and one *n*-alkyl substituent, as examples of such compounds. A study of the influence of members of this library of heterocycles at various concentrations on the surface tension of *m*-xylene, in the accompanying paper [26], revealed considerable surface activity, but indicated some anomalies in the

expected behaviour [24,25]. Similar anomalies of unpredictable behaviour had been reported by others for solutions of polyfluorocarbon-hydrocarbon hybrid molecules in vaseline oil [27]. In this case, it was suggested that the discrepancies were possibly attributable to issues of viscosity. We have chosen to examine the possibility that the unpredictable surface tension behaviour was due to the combination of fluorophilic and lipophilic components of the two quite different sets of heterocyclic and non-heterocyclic compounds. For example, relatively simple and non-polymeric, 1H-1,2,3-triazole derivatives in which there was a perfluorobutylethyl, perfluorohexylethyl, or perfluorooctylethyl substituent at position 1 on the ring, and an *n*-alkyl chain located at position 4 on the ring, showed consistent behaviour when the *n*alkyl chain was 4 or 6 carbons in length, but very different behaviour when the chain was 8 carbons in length. This was irrespective of the fluorous chain length. Our interest therefore turned to analogues of the fluorous heterocyclic molecules in which the partner group was more hydrophilic.

This paper describes the synthesis of fluorous 1,2,3-triazole analogues in which the 1-polyfluoroalkyl group is retained and: (a) the 4-*n*-alkyl chain is replaced by an *n*-alkoxymethyl substituent, **1–3**, or (b) the 4-*n*-alkyl chain is replaced by a methoxytriethyleneglycolmethyl group, **4–6**, and (c) there is a 1-*n*-alkyl substituent in combination with a 4-perfluorohexylethoxymethyl group, **7–11**, or 4-perfluoroheptylmethoxymethyl group, **12–16**, respectively. In addition, a preliminary study towards a new class of ether linked bis-1,2,3-triazoles [14] is described.

^{*} Corresponding author. Tel.: +61 2 9385 4457; fax: +61 2 9385 6190. *E-mail address*: r.read@unsw.edu.au (R.W. Read).

¹ On leave from the Department of Chemistry, University of Central Florida, Orlando, FL 32816-2366, USA.

² On leave from Department of Chemistry, College of Science, Kerbala University, Kerbala, Iraq.

^{0022-1139/\$ –} see front matter \circledcirc 2011 Elsevier B.V. All rights reserved. doi:10.1016/j.jfluchem.2011.07.002

2. Results and discussion

Two synthetic pathways were taken to substances 1-16, and the methods used reflect our modular approach to such fluorous surfactants that are based on relatively simple heterocyclic core molecules [24].

In the first approach, the heterocyclic portion was constructed from fully substituted precursors. Known building blocks, fluorous azides **17–19** [28,29], were made to react with each of propargyl ethers 20 [30] and 21, under Cu(I)-catalysed Huisgen-Meldal 1,3dipolar cycloaddition conditions [31]. A number of different protocols were tested. In our earlier studies [24], aqueous dimethyl sulfoxide [31(d)], was adopted as solvent with good success. It was recognised that dimethyl sulfoxide itself, in the absence of water, was a useful dipolar aprotic solvent for the displacement of iodide in the preparation of perfluoroalkylethyl azides. This prompted a two-step, but one-pot procedure, to be examined in which the azides 17-19 were prepared in situ. Such in situ generation of nonfluorous azides followed by dipolar cycloaddition with copperligated alkynes to generate 1,4-disubstituted triazoles has been reported in various aqueous organic solvent combinations [32-34], but in our case, to respect the hydrophobicity of the fluorous substrates, neat dimethyl sulfoxide was used as solvent in both steps, with no added water. In addition, each step in the overall process using propargyl ether 20 was carried out under microwave irradiation [35], while each step using propargyl ether 21 was carried out under 'conventional' conditions, by regular heating. The former reactions were complete within a total of 3 h while those under conventional conditions required reaction times of 36 h. The reactions all proceeded with excellent regioselectivity to afford the desired 1,4-disubstituted derivatives 1-6 in acceptable to good yields, with the highest derived from the microwave accelerated protocol (Scheme 1). The fluorous azides 17-19 were also prepared separately, purified and reacted directly with propargyl ether 21 under conventional conditions, by way of comparison, but the expected triazoles **4–6** were isolated in only 52–58% yield. When the last of these, reaction between fluorous azide **19** and propargyl ether **21**, was carried out under microwave irradiation conditions, the product was obtained in much higher yield (80%) (Scheme 1). This confirmed the benefits of microwave acceleration, which are well-known, and demonstrated that they applied to this fluorous setting.

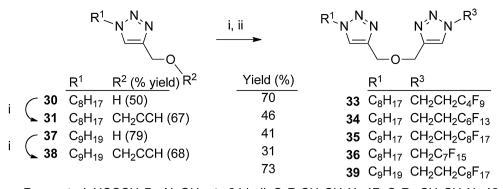
The simpler ether derivatives **1–3** were higher melting than the triethyleneglycol-substituted derivatives 4-6, while the latter were noticeably more difficult to handle and purify.

Next, a series of *n*-alkyl azides, including benzyl azide, **22–26**, were made to react under the same 1,3-dipolar cycloaddition conditions with each of fluorous propargyl ethers, 27 and 28. Compounds 27 and 28 were readily prepared under phase transfer conditions from the corresponding alcohol and propargyl bromide [36]. In this series, the 1,3-dipolar cycloaddition reactions were carried out under conventional conditions, and because the azides 22-26 were prepared from a variety of alkyl halides, they were preprepared separately. Again, the reactions proceeded with virtually perfect regioselectivity to yield the 1,2,3-triazoles 7-16 in good to excellent yields. The compounds were isolated in some cases as oils and waxes and in others crystalline materials. As expected, the more highly fluorinated derivatives, **12–16**, were higher melting than their less fluorinated analogues (see Tables in Supplementary Data).

The influence of the compounds on surface tension of *m*-xylene has been described in the accompanying paper [26], but, by and large, these materials behaved consistently in reducing the surface tension, although again, the 1-n-octvl derivatives showed some deviation from the expected behaviour. Very significantly, it was noted that the hydrophilic derivatives showed higher and more consistent surface activity than the lipophilic analogues. This difference in behaviour was consistent with similar findings for perfluoroalkyl ether-substituted versus perfluoroalkyl-substituted

R ¹ +	$\int_{n} N_3$	+	0-R ²	CuSO ₄ •5H ₂ O	F		_N 	Ň
				$(CH_3)_2SO$, heat				-0 R^2
R ¹	n		R ²	Yield ^a (%)		R ¹	n	R ²
17 C ₄ F ₉	2	20	C ₈ H ₁₇	87 ^{b,c}	1	C_4F_9	2	C ₈ H ₁₇
18 C ₆ F ₁₃	2	21	(CH ₂ CH ₂ O) ₃ CH ₃	81 ^{b,c}	2	$C_{6}F_{13}$	2	C ₈ H ₁₇
19 C ₈ F ₁₇	2	27	CH ₂ CH ₂ C ₆ F ₁₃	83 ^{b,c}	3	C_8F_{17}	2	C ₈ H ₁₇
22 C ₄ H ₉	0	28	$CH_2C_7F_{15}$	52(66 ^b)	4	C_4F_9	2	(CH ₂ CH ₂ O) ₃ CH ₃
23 C ₆ H ₁₃	0	29	Н	58(71 ^b)	5	$C_{6}F_{13}$	2	(CH ₂ CH ₂ O) ₃ CH ₃
24 C ₈ H ₁₇	0			58(80 ^c)(74 ^b)	6	C_8F_{17}	2	(CH ₂ CH ₂ O) ₃ CH ₃
25 C ₉ H ₁₉	0			64	7	C ₄ H ₉	0	$CH_2CH_2C_6F_{13}$
26 CH ₂ Ph	0			97	8	$C_{6}H_{13}$	0	CH ₂ CH ₂ C ₆ F ₁₃
				90	9	C_8H_{17}	0	CH ₂ CH ₂ C ₆ F ₁₃
				82	10	C_9H_{19}	0	$CH_2CH_2C_6F_{13}$
NOTES				88	11	CH_2Ph	0	CH ₂ CH ₂ C ₆ F ₁₃
^a Conventional reaction conditions				84	12	C₄H ₉	0	CH ₂ C ₇ F ₁₅
(see Experimental for details).				88	13	C_6H_{13}	0	$CH_2C_7F_{15}$
^b In situ preparation of azide.				82	14	C_8H_{17}	0	CH ₂ C ₇ F ₁₅
^c Microwave accelerated conditions.				81	15	C_9H_{19}	0	$CH_2C_7F_{15}$
^d Reaction performed in aqueous				74	16	CH_2Ph	0	CH ₂ C ₇ F ₁₅
<i>tert</i> -BuOH instead of $(CH_3)_2SO$.				50 ^d	30	C ₈ H ₁₇	0	Н

Scheme 1. Preparation of triazolyl ethers 1-16 and hydroxymethyltriazole 30 through Cu(I)-catalysed azide-alkyne dipolar cycloaddition.



Reagents: i. HCCCH₂Br, NaOH, r.t., 24 h; ii. $C_4F_9CH_2CH_2N_3$ **17**, $C_6F_{13}CH_2CH_2N_3$ **18**, $C_8F_{17}CH_2CH_2N_3$ **19**, or $C_7F_{15}CH_2N_3$ **32**, and $CuSO_4 \cdot 5H_2O$, Na ascorbate, $(CH_3)_2SO$, 85°C, 48 h.

Scheme 2. Preparation of hydroxymethyltriazoles 30 and 37, triazolyl propargyl ethers 31 and 38, and di(triazolylmethyl) ethers 33-36 and 39.

benzenes [37]. An alternative approach to ether-substituted triazoles involving functionalisation of preformed heterocycles was therefore investigated and access to more elaborate fluorous 1,2,3-triazole derivatives was also explored.

There has been some interest in gemini fluorous surfactants, such as bis-tetraalkylammonium salts [19], because they have enhanced bioactivity and can form unique self assembled clusters. The fluorous, ether-substituted triazoles have similar potential to be fashioned into interesting gemini heterocycles through synthesis of ether linked di-heterocycles. We wished to demonstrate this concept in principle and have synthesised two pseudo symmetrical and three unsymmetrical examples. Huisgen-Meldal dipolar cycloaddition [32] between *n*-octyl azide **25** and propargyl alcohol 29 proceeded well in aqueous tert-butanol (Scheme 1) and the intermediate 4-hydroxymethyl-1-n-octyl-1,2,3-triazole 30 [35] was treated with propargyl bromide under phase transfer conditions with sodium hydroxide as base (Scheme 2). Subsequent reaction of the propargyl ether **31** with perfluorobutylethyl azide 17, perfluorohexylethyl azide 18, perfluorooctylethyl azide 19, and perfluoroheptylmethyl azide 32, again under copper(I)-promoted click conditions [31], gave the desired, new bis-1,2,3-triazoles 33-36. Similarly, dipolar cycloaddition of *n*-nonyl azide 25 with propargyl alcohol 29 yielded new 4-hydroxymethyl-1-n-nonyl-1,2,3-triazole 37, which was subsequently transformed into its propargyl ether 38. Treatment of propargyl ether 38 with perfluorooctylethyl azide 19 under copper(I)-promoted conditions proceeded without incident to bis-1,2,3-triazole 39.

The structures of triazoles **33–36** contain the same *n*-octyl substituent in one triazole ring but (a) increasing fluorous substituents of increasing chain length (**33–35**), and (b) fluorous chains of equal length and increasing fluorine content (**34** and **36**), in the second ring. Meanwhile, bis-triazoles **35** and **39** bear identical N-perfluorooctylethyl substituents in one heterocyclic ring but non-fluorous *n*-alkyl substituents of increasing chain length in the other. Together, this collection of new compounds comprises a valuable small library with highly systematic structural variations for further study.

The scope of these functionalised fluorous molecules for selfassembly, drug transport, and additional functionalisation, will be investigated in future.

3. Experimental

3.1. General

Melting points were determined using a Köfler hotstage melting point apparatus and are uncorrected. Infrared spectra were recorded on a Nicolet Avatar 320 FT-IR spectrophotometer as thin films of neat liquid and in KBr disks for solids; measurements are reported as wavenumbers in cm⁻¹. Routine ¹H, ¹³C and ¹⁹F NMR spectra were obtained on a Bruker DPX300F spectrometer. Chemical shifts were recorded in parts per million (ppm) relative to solvent nuclei as an internal reference. Mass spectra were measured using the electrospray ionization (ESI) technique, at low resolution on a Waters Micromass ZQ2000 LC–MS instrument by direct injection and at high resolution using a Thermo LTQ Orbitrap XL instrument. Samples were dissolved in MeOH. Microanalyses were carried out at the Microanalytical Laboratory, Australian National University. All chemicals were commercial reagent grade unless otherwise specified.

3.2. General method for preparation of alkynyl ethers 20 and 21

A dry flask was charged with THF (15 mL), NaOH pellets (2.40 g, 60 mmol), and *n*-octanol or triethyleneglycol monomethyl ether (15 mmol). The flask was immersed in an ice-bath and the contents stirred for 10 min. before propargyl bromide (2.51 mL, 22.5 mmol) was added dropwise. The reaction mixture was then allowed to stir for 24 h, gradually warming to r.t. The reaction mixture was poured into Et_2O (20 mL) and water (50 mL) and the layers were separated. The aqueous layer was further extracted with Et_2O (2 × 20 mL), and the combined organic extracts were washed with 10% aq. HCl, followed by sat. NaHCO₃ solution and brine, and then dried over MgSO₄. Column chromatography (silica gel, Et_2O /light petroleum, 10:90) followed by Kulgelrohr distillation gave the desired products.

3.2.1. 1-(Prop-2-ynyloxy)octane 20

n-Octanol (1.95 g) gave 1-(prop-2-ynyloxy)octane **20** as a colourless oil (2.39 g, 95%) bp 79–81 °C/0.1 mmHg (lit. [30]: bp 80 °C (oven)/0.13 mmHg). IR (neat): 3583, 3312, 2927, 2856, 2117, 1467, 1443, 1378, 1357, 1269, 1104, 1026, 945, 917, 722, 665, 622, 422 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 0.87 (t, *J* 6.6 Hz, 3H, CH₃CH₂), 1.26 (m, 8H, (CH₂)₄CH₃), 1.33 (m, 2H, CH₂CH₂CH₂O), 1.58 (m, 2H, CH₂CH₂O), 2.40 (t, *J* 2.4 Hz, 1H, OCH₂CCH), 3.49 (t, *J* 6.7 Hz, 2H, CH₂CH₂O), 4.12 (d, *J* 2.4 Hz, 2H, OCH₂CCH). ¹³C NMR (75 MHz, CDCl₃) δ : 14.0 (CH₃), 22.5 (CH₂CH₃), 26.0 (CH₂), 29.1 (CH₂), 29.4 (CH₂), 29.6 (CH₂), 31.7 (CH₂CH₂O), 57.9 (OCH₂CCH), 70.2 (CH₂OCH₂CCH), 73.9 (OCH₂CCH), 80.0 (OCH₂CCH). HR-MS (ESI) Calcd. for [C₁₁H₂₀O+H]⁺: *m*/z 169.1587. Found: *m*/z 169.1582. Anal. Calcd. for C₁₁H₂₀O·0.33H₂O: C, 75.83; H, 11.96%. Found: C, 75.73; H, 12.11%.

3.2.2. 3-(2-(2-(2-Methoxy)ethoxy)ethoxy)prop-1-yne 21

Triethyleneglycol monomethyl ether (2.46 g) gave 3-(2-(2-(2-methoxy)ethoxy)ptop-1-yne **21** as a colourless oil

(2.48 g, 82%) bp 130–133 °C (oven)/0.1 mmHg. IR (neat): 3582, 3428, 3278, 2880, 2114, 1724, 1650, 1454, 1352, 1287, 1247, 1197, 1104, 1032, 935, 887, 850, 666 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 2.41 (t, *J* 2.4 Hz, 1H, OCH₂CCH), 3.36 (s, 3H, OCH₃), 3.54 (m, 2H, CH₂CH₂OCH₃), 3.62 (m, 2H, OCH₂CH₂OCH₃), 3.65 (m, 4H, 2 × OCH₂CH₂), 3.66 (m, 2H, CH₂OCH₂CCH), 3.67 (m, 2H, OCH₂CH₂OCH), 4.18 (d, *J* 2.4 Hz, 2H, OCH₂CCH). ¹³C NMR (75 MHz, CDCl₃) δ : 58.3 (OCH₂CCH), 58.9 (OCH₃), 69.0 (CH₂OCH₂CCH), 70.3 (OCH₂CH₂), 70.4 (OCH₂CH₂), 70.5 (2 × OCH₂CH₂), 71.8 (CH₂OCH₃), 74.4 (OCH₂CCH), 79.5 (OCH₂CCH). MS (ESI) *m/z*: 226.17 ([M+H+Na]⁺, 10%), 225.16 ([M+Na]⁺, 100). Anal. Calcd. for C₁₀H₁₈O₄ requires C, 59.39; H, 8.97%. Found: C, 59.24; H, 9.28%.

3.3. General method for the microwave accelerated reactions of propargyl ether 20 with in situ prepared perfluoroalkylethyl azides 17–19

Perfluoroalkylethyl iodide (1.0 mol equiv.) (3-4 mmol) and NaN₃ (1.2 mol equiv.) were added to dry DMSO (2 mL) in a 10 mL CEM microwave glass reactor tube, a magnetic stirring bead added and the system sealed. The tube was subjected to stirring and microwave irradiation at 250 W and 65 °C for 1 h then immediately cooled to r.t. The tube was opened and *n*-octyl propargyl ether 41 (1.0 mol equiv.) added, followed by sodium ascorbate (0.1 mol equiv.) and CuSO₄·5H₂O (0.05 mol equiv.) and additional DMSO (1 mL). The tube was again sealed and again subjected to stirring and microwave irradiation at elevated temperature. At the conclusion of the reaction, the tube was cooled, opened, and the contents diluted with water (25 mL). The product was then extracted into Et_2O (3 × 20 mL), and the combined extracts washed in turn with brine $(2 \times 10 \text{ mL})$, H₂O $(2 \times 10 \text{ mL})$ and again brine (10 mL), and then dried and evaporated. The residue was chromatographed on silica gel and the product eluted with a gradient of Et₂O/light petroleum.

3.3.1. 4-((Octyloxy)methyl)-1-(2-(perfluorobutyl)ethyl)-1H-1,2,3triazole 1

Perfluorobutylethyl iodide (1.496 g, 4.0 mmol) was converted to its azide **17** at 75 °C for 1 h and then treated with propargyl ether **20** (0.673 g, 4.0 mmol) at 95 °C for 2 h to afford 4-((octyloxy)methyl)-1-(2-(perfluorobutyl)ethyl)-1H-1,2,3-triazole **1** as white needles (1.59 g, 87%) mp 40-42 °C (Et₂O-light petroleum). IR (KBr): 3500, 2924, 1467, 1359, 1225, 1134, 1053, 1018, 988, 881, 859, 771, 738, 700, 668, 646, 596, 531 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) *δ*: 0.86 (t, *J* 7.0 Hz, 3H, CH₃), 1.28 (m, 8H, (CH₂)₄CH₃), 1.31 (m, 2H, OCH₂CH₂CH₂), 1.58 (tt, J 7.4, 6.7 Hz, 2H, OCH₂CH₂), 2.81 (tt, ³*J*_{H-F} 18.0, ³*J*_{H-H} 7.4 Hz, 2H, CH₂CF₂), 3.50 (t, *J* 6.7 Hz, 2H, OCH₂CH₂), 4.61 (s, 2H, 4-CH₂O), 4.66 (dd, J 7.6, 7.3 Hz, 2H, N1CH₂CH₂CF₂), 7.58 (s, 1H, H5). ¹³C NMR (100 MHz, CDCl₃) δ: 14.2 (CH₃), 22.8 (CH₂CH₃), 26.2 (CH₂), 29.4 (CH₂), 29.5 (CH₂), 29.8 (CH₂), 31.90 (t, ²J_{C-F} 21.7 Hz, N1CH₂CH₂CF₂), 31.93 (OCH₂CH₂), 42.4 (br t, ${}^{3}J_{C-F}$ 5.2 Hz, N1CH₂CH₂CF₂), 64.4 (4-CH₂O), 71.2 (OCH₂CH₂), 122.9 (C5), 146.2 (C4). MS (ESI) m/z: 481.07 ([M+H+Na]⁺, 31%), 480.13 ([M+Na]⁺, 100). Anal. Calcd. for C₁₇H₂₄F₉N₃O requires: C, 44.64; H, 5.29; N, 9.19%. Found: C, 44.76; H, 5.02; N, 9.33%.

3.3.2. 4-((Octyloxy)methyl)-1-(2-(perfluorohexyl)ethyl)-1H-1,2,3-triazole 2

Perfluorohexylethyl iodide (1.422 g, 3.0 mmol) was converted to its azide **18** through treatment at 75 °C for 1 h, then treated with propargyl ether **20** (0.505 g, 3.0 mmol) at 95 °C for 2 h to afford 4- *((octyloxy)methyl)-1-(2-(perfluorohexyl)ethyl)-1H-1,2,3-triazole* **2** as white needles (1.35 g, 81%) mp 52–54 °C (Et₂O-light petroleum). IR (KBr): 3521, 3101, 2935, 1468, 1240, 1196, 1138, 1118, 1095, 1078, 1054, 998, 982, 845, 786, 768 725, 712, 680, 642, 565 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ : 0.85 (t, J 6.8 Hz, 3H, CH₃), 1.24 (m, 8H,

 $(CH_2)_4CH_3$), 1.31 (m, 2H, OCH₂CH₂CH₂), 1.58 (tt, *J* 7.0, 6.6 Hz, 2H, OCH₂CH₂), 2.80 (tt, ³*J*_{H-F} 18.0, ³*J*_{H-H} 7.4 Hz, 2H, CH₂CF₂), 3.50 (t, *J* 6.6 Hz, 2H, OCH₂CH₂), 4.62 (s, 2H, 4-CH₂O), 4.67 (dd, *J* 7.5, 7.4 Hz, 2H, N1CH₂CH₂CF₂), 7.61 (s, 1H, H5). ¹³C NMR (100 MHz, CDCl₃) δ : 14.2 (CH₃), 22.8 (CH₂CH₃), 26.2 (CH₂), 29.4 (CH₂), 29.5 (CH₂), 29.7 (CH₂), 31.8 (t, ²*J*_{C-F} 21.7 Hz, N1CH₂CH₂CF₂), 31.9 (OCH₂CH₂), 42.2 (br t, ³*J*_{C-F} 4.9 Hz, N1CH₂CH₂CF₂), 64.3 (4-CH₂O), 71.1 (OCH₂CH₂), 123.1 (C5-H), 146.0 (C4). MS (ESI) *m/z*: 596.00 ([M+K]⁺, 3%), 581.03 ([M+H+Na]⁺, 25), 580.03 ([M+Na]⁺, 100), 558.04 ([M+H]⁺, 14). Anal. Calcd. for C₁₉H₂₄F₁₃N₃O requires: C, 40.94; H, 4.34; N, 7.54%. Found: C, 40.71; H, 4.46; N, 7.77%.

3.3.3. 4-((Octyloxy)methyl)-1-(2-(perfluorooctyl)ethyl)-1H-1,2,3triazole 3

Perfluorooctylethyl iodide (1.722 g, 3.0 mmol) was converted to its azide **19** through microwave irradiation at 85 °C for 1 h, then treated with propargyl ether 20 (0.505 g, 3.0 mmol) at 98 °C for 2.5 h to afford 4-((octyloxy)methyl)-1-(2-(perfluorooctyl)ethyl)-1H-1,2,3-triazole **3** as white needles (1.63 g, 83%) mp 80-82 °C (Et₂Olight petroleum). IR (KBr): 3646, 3410, 2933, 1443, 1359, 1204, 1148, 1116, 1063, 1032, 983, 957, 921, 746, 706, 660, 585, 557, 518 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ: 0.86 (t, J 6.8 Hz, 3H, CH₃), 1.24 (m, 8H, (CH₂)₄CH₃), 1.31 (m, 2H, OCH₂CH₂CH₂), 1.58 (tt, J 7.1, 6.8 Hz, 2H, OCH₂CH₂), 2.81 (tt, ³J_{H-F} 18.0, ³J_{H-H} 7.5 Hz, 2H, CH₂CF₂), 3.50 (t, J 6.7 Hz, 2H, OCH₂CH₂), 4.62 (s, 2H, 4-CH₂O), 4.68 (dd, J 7.5, 7.3 Hz, 2H, N1CH₂CH₂CF₂), 7.61 (s, 1H, H5). ¹³C NMR (100 MHz, CDCl₃) δ: 14.2 (CH₃), 22.8 (CH₂CH₃), 26.2 (CH₂), 29.4 (CH₂), 29.5 (CH₂), 29.7 (CH₂), 31.86 (t, ²J_{C-F} 21.8 Hz, N1CH₂CH₂CF₂), 31.93 (OCH₂CH₂), 42.4 (br t, ³J_{C-F} 4.6 Hz, N1CH₂CH₂CF₂), 64.3 (4-CH₂O), 71.1 (OCH₂CH₂), 123.1 (C5-H), 146.1 (C4). MS (ESI) m/z: 680.93 ([M+Na]⁺, 23%), 679.93 ([M-H+Na]⁺, 100). Anal. Calcd. for C₂₁H₂₄F₁₇N₃O requires: C, 38.37; H, 3.68; N, 6.39%. Found: C, 38.44; H, 3.76; N, 6.46%.

3.4. General method for the preparation of 4-((2-(2-(2methoxyethoxy)ethoxy)methyl)-1-(2-perfluoroalkylethyl)-1H-1,2,3-triazoles 4–6 by conventional, thermal reactions of propargyl ether 21 with in situ prepared perfluoroalkylethyl azides 17–19

Perfluoroalkylethyl iodide (1.0 mol equiv.) (3–5 mmol) and NaN₃ (1.0–1.5 mol equiv.) were added to dry DMSO (4 mL), a magnetic stirring bead added, and the mixture stirred under argon in a bath at 65 °C for 24 h then cooled to r.t. The methoxytriethylenoxy propargyl ether **21** (1.0 mol equiv.) added, followed by sodium ascorbate (0.1 mol equiv.) and $CuSO_4$ ·5H₂O (0.05 mol equiv.), and additional DMSO (2 mL). The mixture was again stirred under argon at elevated temperature for 48 h. The contents were cooled and diluted with water (50 mL). The product was then extracted into Et₂O (3 × 20 mL), and the combined extracts washed in turn with brine (2 × 10 mL), H₂O (2 × 10 mL) and again brine (10 mL), then dried and evaporated. The residue was chromatographed on silica gel and the product eluted with a gradient of Et₂O/light petroleum and distilled or characterized directly as an oil.

3.4.1. 4-((2-(2-(2-Methoxyethoxy)ethoxy)ethoxy)methyl)-1-(2perfluorobutylethyl)-1H-1,2,3-triazole 4

Perfluorobutylethyl iodide (1.87 g, 5.0 mmol) was treated with NaN₃ (0.49 g, 7.5 mmol) at 65 °C for 24 h to generate azide **17**, which was then treated with propargyl ether **21** (1.02 g, 5.0 mmol) at 90 °C for 48 h to give 4-((2-(2-(2-methoxyethoxy)ethoxy)ethoxy)methyl)-1-(2-perfluorobutylethyl)-1H-1,2,3-triazole **4** as pale yellow wax (1.62 g, 66%) bp 223–228 °C (oven)/0.1 mmHg. IR (neat): 3139, 2882, 2357, 2248, 1714, 1664, 1646, 1455, 1403, 1352, 1223, 1135, 1049, 990, 928, 881, 858, 832, 748, 710, 666 cm⁻¹. ¹H NMR

(300 MHz, CDCl₃) δ: 2.78 (tt, ${}^{3}J_{H-F}$ 18.1, ${}^{3}J_{H-H}$ 7.2 Hz, 2H, $CH_{2}CF_{2}$), 3.32 (s, 3H, OCH₃), 3.50 (t, *J* 4.4 Hz, 2H, $CH_{2}OCH_{3}$), 3.55–3.67 (m, 10H, 5 × OCH₂CH₂), 4.64 (partially obscured dd, *J* 7.2, *ca*. 6.4 Hz, 2H, N1CH₂CH₂CF₂), 4.65 (s, 2H, 4-CH₂O), 7.65 (s, 1H, H5). ${}^{13}C$ NMR (75 MHz, CDCl₃) δ: 31.6 (t, ${}^{2}J_{C-F}$ 21.7 Hz, N1CH₂CH₂CF₂), 42.1 (t, ${}^{3}J_{C-F}$ *ca*. 4 Hz, N1CH₂CH₂CF₂), 58.8 (OCH₃), 64.5 (4-CH₂O), 69.7 (4-CH₂OCH₂), 70.35 (OCH₂CH₂), 70.40 (OCH₂CH₂), 70.42 (OCH₂CH₂), 70.47 (OCH₂CH₂), 71.8 (CH₂OCH₃), 123.0 (C5-H), 145.6 (C4). MS (ESI) *m/z*: 515.29 ([M+H+Na]⁺, 23%), 514.22 ([M+Na]⁺, 100), 492.10 ([M+H]⁺, 20). HR-MS (ESI) Anal. Calcd. for [C₁₆H₂₂N₃O₄F₉+Na+H]⁺ *m/z*: 515.1443. Found: *m/z* 515.1390 (19%); Anal. Calcd. for [C₁₆H₂₂N₃O₄F₉+Na]⁺ *m/z* 514.1365. Found: *m/z* 514.1361 (100); Anal. Calcd. for [C₁₆H₂₂N₃O₄F₉+H]⁺ *m/z* 492.1545. Found: *m/z* 492.1549 (15).

3.4.2. 4-((2-(2-(2-Methoxyethoxy)ethoxy)ethoxy)methyl)-1-(2-perfluorohexylethyl)-1H-1,2,3-triazole 5

Perfluorohexylethyl iodide (2.37 g, 5.0 mmol) was treated with NaN₃ (0.48 g, 7.5 mmol) at 65 °C for 24 h to generate azide 18, which was then treated with propargyl ether 21 (1.03 g, 5.1 mmol) in a bath at 100 °C for 48 h to give 4-((2-(2-(2methoxyethoxy)ethoxy)methyl)-1-(2-perfluorohexylethyl)-1H-1,2,3-triazole **5** as a pale yellow wax (2.09 g, 71%). IR (neat): 3138, 2879, 1731, 1646, 1456, 1402, 1366, 1351, 1319, 1240, 1206, 1145, 1122, 1050, 989, 949, 850, 810, 746, 736, 708, 699, 666 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ: 2.82 (tt, ³J_{H-F} 18.5, ³J_{H-H} 7.2 Hz, 2H, CH₂CF₂), 3.37 (s, 3H, OCH₃), 3.55 (t, J 4.7 Hz, 2H, CH_2OCH_3), 3.60–3.80 (m, 10H, 5 × OCH_2CH_2), 4.68 (dd, J 7.5, 7.5 Hz, 2H, N1CH₂CH₂CF₂), 4.71 (s, 2H, 4-CH₂O), 7.71 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ: 31.7 (t, ²*J*_{C-F} 21.7 Hz, N1CH₂CH₂CF₂), 42.1 (br, N1CH₂CH₂CF₂), 58.8 (OCH₃), 64.5 (4-CH₂O), 69.7 (4-CH₂OCH₂), 70.35 (OCH₂CH₂), 70.40 (OCH₂CH₂), 70.42 (OCH₂CH₂), 70.48 (OCH₂CH₂), 71.8 (CH₂OCH₃), 123.1 (C5-H), 145.7 (C4). MS (ESI) m/z: 615.36 ([M+H+Na]⁺, 8%), 614.30 ([M+Na]⁺, 100). HR-MS (ESI) Anal. Calcd. for $[C_{18}H_{22}N_3O_4F_{13}+Na+H]^+ m/z$: 615.1379. Found: m/z 615.1337 (21%); Anal. Calcd. for $[C_{18}H_{22}N_3O_4F_{13}+Na]^+$ requires *m*/*z* 614.1301. Found: *m*/*z* 614.1309 (100). Anal. Calcd. for $[C_{18}H_{22}N_3O_4F_{13}+H]^+$ requires m/z 592.1481. Found: m/z592.1500 (22).

3.4.3. 4-((2-(2-(2-Methoxyethoxy)ethoxy)ethoxy)methyl)-1-(2-perfluorooctylethyl)-1H-1,2,3-triazole 6

Perfluorooctylethyl iodide (1.72 g, 3.0 mmol) was treated with NaN₃ (0.23 g, 3.0 mmol) in a bath at 70 °C for 24 h to generate azide **19**, which was then treated with propargyl ether **21** (0.61 g, 3.0 mmol) in a bath at 110 °C for 48 h to give 4-((2-(2-(2methoxyethoxy)ethoxy)methyl)-1-(2-perfluorooctylethyl)-1H-1,2,3-triazole 6 as a white powder (1.53 g, 79%) mp 47-49 °C (Et₂O/light petroleum). IR (KBr): 3126, 2875, 2368, 1715, 1557, 1539, 1455, 1401, 1372, 1338, 1201, 1151, 1115, 1053, 989, 958, 851, 706, 660 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 2.82 (tt, ³ J_{H-F} 18.5, ³J_{H-H} 7.5 Hz, 2H, CH₂CF₂), 3.36 (s, 3H, OCH₃), 3.55 (t, J 4.2 Hz, 2H, CH₂OCH₃), 3.60–3.75 (m, 10H, 5 × OCH₂CH₂), 4.68 (dd, J 7.5, 7.5 Hz, 2H, N1CH₂CH₂CF₂), 4.71 (s, 2H, 4-CH₂O), 7.70 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ: 31.6 (t, ²J_{C-F} 21.7 Hz, N1CH₂CH₂CF₂), 42.1 (br, N1CH₂CH₂CF₂), 58.7 (OCH₃), 64.4 (4-CH₂O), 69.6 (4-CH₂OCH₂), 70.30 (OCH₂CH₂), 70.36 (OCH₂CH₂), 70.38 (OCH₂CH₂), 70.43 (OCH₂CH₂), 71.7 (CH₂OCH₃), 123.2 (C5-H), 145.5 (C4). MS (ESI) *m*/*z*: 715.36 ([M+H+Na]⁺, 14%), 714.23 ([M+Na]⁺, 100). Anal. Calcd. for C₂₀H₂₂F₁₇N₃O₄ requires: C, 34.74; H, 3.21; N, 6.08%. Found: C, 34.63; H, 2.97; N, 5.90%.

3.5. General method for the preparation of alkynyl ethers 27 and 28

A dry flask was charged with THF (9 mL), NaOH pellets (0.48 g, 12 mmol), and perfluorohexylethanol or perfluoroheptylmethanol

(3 mmol). The flask was immersed in an ice-bath and the contents stirred for 10 min. Propargyl bromide (0.54 g, 4.5 mmol) was added dropwise and the reaction mixture stirred and gradually allowed to warm to r.t. over 24 h. The reaction mixture was poured into Et₂O (20 mL) and water (50 mL), the layers separated and the aqueous layer extracted with more Et₂O (3×20 mL). The combined organic extracts were washed with 10% HCl soln. followed by sat. aq. NaHCO₃ and brine, and then dried over Na₂SO₄. Column chromatography (silica gel, Et₂O/LP, 10:90) followed by Kulgelrohr distillation gave the desired products.

3.5.1. 3-((2-Perfluorohexyl)ethoxy)prop-1-yne 27

Perfluorohexylethanol (1.09 g) gave 3-((2-perfluorohexyl)ethoxy)prop-1-yne **27** as a colourless oil (0.96 g, 80%) bp 125–128 °C (oven)/0.1 mmHg. IR (neat): 3850, 3608, 3683, 3317, 2960, 2898, 2859, 2360, 2342, 2122, 1362, 1240, 1146, 1010, 913, 811, 708, 698, 653 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 2.43 (ttt, ³J_{H-F} 18.6, ³J_{H-H} 6.8, ⁴J_{H-F} 1.4 Hz, 2H, OCH₂CH₂CF₂), 2.46 (t, *J* 2.4 Hz, 1H, OCH₂CCH), 3.82 (t, *J* 6.8 Hz, 2H, OCH₂CH₂CF₂), 4.18 (d, *J* 2.4 Hz, 2H, OCH₂CCH). ¹³C NMR (75 MHz, CDCl₃) δ : 31.3 (t, ²J_{C-F} 21.5 Hz, OCH₂CH₂CF₂), 58.3 (OCH₂CCH), 61.6 (t, ³J_{C-F} ca. 5 Hz, OCH₂CH₂CF₂), 74.9 (OCH₂CCH), 78.8 (OCH₂CCH). MS (ESI) *m*/*z*: 425.14 ([M+Na]⁺). Anal. Calcd. for C₁₁H₇F₁₃O requires C, 32.85; H, 1.75%. Found: C, 32.53; H, 1.68%.

3.5.2. 3-((2-Perfluoroheptyl)methoxy)prop-1-yne 28

Perfluoroheptylmethanol (1.20 g) gave 3-(*perfluoroheptylmethoxy*)*prop-1-yne* **28** as a colourless oil (1.12 g, 85%) b.p. 130–132 °C (oven)/0.1 mmHg. IR (neat): 3850, 3583, 3317, 2924, 2360, 2342, 2126, 1690, 1464, 1448, 1362, 1328, 1213, 1150, 1090, 1023, 963, 884, 808, 722, 702, 660 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 2.54 (t, *J* 2.4 Hz, 1H, OCH₂CCH), 4.05 (tt, ³*J*_{H-F} 13.8, ⁴*J*_{H-F} 1.5 Hz, 2H, OCH₂CF₂), 4.32 (d, *J* 2.4 Hz, 2H, OCH₂CCH). ¹³C NMR (75 MHz, CDCl₃) δ : 59.3 (OCH₂CCH), 65.7 (t, *J* 25.5 Hz, OCH₂CF₂), 76.2 (OCH₂CCH), 77.4 (OCH₂CCH). MS (ESI) *m/z*: 439.21([M+1]⁺). Anal. Calcd. for C₁₁H₅F₁₅O requires C, 30.16; H, 1.15%. Found: C, 30.32; H, 1.08%.

3.6. General procedure for the preparation of triazoles 7–16

Propargyl ether (alkyne) (1 mmol) and alkyl azide (2 mmol) were added to a suspension of sodium ascorbate (0.0198 g, 0.10 mmol) and CuSO₄·5H₂O (0.0125 g, 0.05 mmol) in (CH₃)₂SO (4 mL). The mixture was stirred and heated in a bath to 60 °C for 48 h. The reaction mixture was diluted with water (20 mL), extracted with Et₂O (3×15 mL). The combined organic layers were washed with brine (2×10 mL), dried over Na₂SO₄, and evaporated to dryness under reduced pressure. The residue was flash chromatographed (silica gel, Et₂O/LP 10:90 to 20:80) and the main fraction recrystallized from light petroleum where appropriate.

3.6.1. Preparations involving 3-(2-perfluorohexylethoxy)propyne 27 Propargyl ether **27** (0.40 g, 1.00 mmol) was made to react with the following sections.

3.6.1.1. 1-n-Butyl-4-(2-perfluorohexylethoxy)methyl-1H-1,2,3-triazole 7. n-Butyl azide (0.20 g, 2.02 mmol) to give 1-n-butyl-4-(2perfluorohexylethoxy)methyl-1H-1,2,3-triazole **7** as a pale yellow oil (0.32 g, 64%). IR (neat): 3850, 3583, 3444, 3137, 2965, 2939, 2879, 2359, 2342, 1643, 1468, 1441, 1366, 1318, 1240, 1206, 1146, 1124, 1049, 1023, 951, 810, 708, 698, 652 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 0.95 (t, J 7.5 Hz, 3H, CH₃), 1.36 (qt, J 7.5, 7.2 Hz, 2H, CH₂CH₃), 1.89 (tt, J 7.5, 7.2 Hz, 2H, N1CH₂CH₂), 2.42 (tt, ³J_{H-F} 18.5, ³J_{H-H} 6.4 Hz, 2H, OCH₂CH₂CF₂), 3.82 (t, J 6.4 Hz, 2H, OCH₂CH₂CF₂), 4.36 (t, J 7.5 Hz, 2H, N1CH₂CH₂), 4.68 (s, 2H, 4-CH₂O), 7.52 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ: 13.2 (CH₃), 19.6 (CH₂CH₃), 31.4 (t, ${}^{2}J_{C-F}$ 22.4 Hz, OCH₂CH₂CF₂), 32.1 (N1CH₂CH₂), 50.0 (N1CH₂CH₂), 62.1 (br, OCH₂CH₂CF₂), 64.5 (4-CH₂O), 122.2 (C5-H), 144.4 (C4). MS (ESI) *m/z*: 540.33 ([M+K]⁺), 524.31 ([M+Na]⁺). Anal. Calcd. for C₁₅H₁₆F₁₃N₃O requires: C, 35.94; H, 3.22; N, 8.38%. Found: C, 35.95; H, 3.33; N, 8.12%.

3.6.1.2. 1-n-Hexvl-4-(2-perfluorohexvlethoxv)methvl-1H-1.2.3-triazole 8. *n*-Hexyl azide (0.25 g, 1.97 mmol) to give 1-*n*-hexyl-4-(2perfluorohexylethoxy)methyl-1H-1,2,3-triazole 8 as pale yellow wax (0.51 g, 97%) mp 31-32 °C. IR (KBr): 3130, 2960, 2935, 2862, 2362, 1466, 1445, 1368, 1320, 1238, 1208, 1192, 1145, 1124, 1107, 1091, 1057, 1008, 951, 844, 708, 700, 653 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 0.88 (t, / 6.8 Hz, 3H, CH₃), 1.27 (m, 2H, CH₂CH₃), 1.31 (m, 4H, N1CH₂CH₂(CH₂)₂), 1.90 (tt, J 6.8, 6.8 Hz, 2H, N1CH₂CH₂), 2.42 (tt, ³*J*_{H-F} 18.8, ³*J*_{H-H} 6.8 Hz, 2H, OCH₂CH₂CF₂), 3.82 (t, *J* 6.8 Hz, 2H, OCH₂CH₂CF₂), 4.34 (t, J 7.5 Hz, 2H, N1CH₂CH₂), 4.67 (s, 2H, 4-CH₂O), 7.51 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ: 13.7 (CH₃), 22.3 (CH₂CH₃), 26.0 (CH₂), 30.1 (CH₂), 31.0 (N1CH₂CH₂), 31.4 (t, ²J_C-F 21.7 Hz, OCH₂CH₂CF₂), 50.3 (N1CH₂CH₂), 62.1 (br, OCH₂CH₂CF₂), 64.5 (4-CH₂O), 122.2 (C5-H), 144.4 (C4). MS (ESI), m/z: 553.48 ([M+H+Na]⁺, 30%), 552.53 ([M+Na]⁺, 100). Anal. Calcd. for C17H20F13N3O requires: C, 38.57; H, 3.81; N, 7.94%. Found: C, 38.95; H, 3.83; N, 7.72%.

3.6.1.3. 1-n-Octyl-4-(2-perfluorohexylethoxy)methyl-1H-1,2,3-triazole 9. n-Octyl azide (0.31 g, 2.00 mmol) to give 1-n-octyl-4-(2perfluorohexylethoxy)methyl)-1H-1,2,3-triazole **9** as white powder (0.50 g, 90%) mp 34–35 °C. IR (KBr): 3132, 2958, 2928, 2856, 2360, 1542, 1466, 1368, 1322, 1237, 1192, 1144, 1124, 1109, 1092, 1057, 1006, 951, 835, 789, 731, 702, 654 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ: 0.87 (t, / 6.8 Hz, 3H, CH₃), 1.26 (m, 6H, (CH₂)₃CH₃), 1.32 (m, 4H, N1CH₂CH₂(CH₂)₂), 1.90 (tt, 17.2, 6.8 Hz, 2H, N1CH₂CH₂), 2.42 (tt, ³/_{H-F} 18.8, ³/_{H-H} 6.8 Hz, 2H, OCH₂CH₂CF₂), 3.82 (t, / 6.8 Hz, 2H, OCH₂CH₂CF₂), 4.34 (t, 17.5 Hz, 2H, N1CH₂CH₂), 4.67 (s, 2H, 4-CH₂O), 7.51 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ: 13.8 (CH₃), 22.4 (CH₂CH₃), 26.3 (CH₂), 28.8 (CH₂), 30.2 (CH₂), 31.4 (t, ²J_{C-F} 21.3 Hz, OCH₂CH₂CF₂), 31.7 (N1CH₂CH₂), 50.3 (N1CH₂CH₂), 62.1 (t, J 4.3 Hz, OCH₂CH₂CF₂), 64.5 (4-CH₂O), 122.2 (C5-H), 144.4 (C4). MS (ESI) *m*/*z*: 580.59 ([M+Na]⁺, 100%). Anal. Calcd. for C₁₉H₂₄F₁₃N₃O requires: C, 40.94; H, 4.59; N, 7.54%. Found: C, 41.25; H, 4.53; N, 7.29%.

3.6.1.4. 1-n-Nonyl-4-(2-perfluorohexylethoxy)methyl-1H-1,2,3-triazole 10. n-Nonyl azide (0.58 g, 3.43 mmol) to give 1-n-nonyl-4-(2perfluorohexylethoxy)methyl)-1H-1,2,3-triazole 10 as white powder (0.47 g, 82%) mp 41–43 °C. IR (KBr): 3842, 3133, 2957, 2925, 2851, 2360, 2341, 1547, 1467, 1368, 1322, 1236, 1207, 1191, 1144, 1124, 1109, 1056, 1028, 843, 790, 732, 702, 654 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ: 0.87 (t, J 6.4 Hz, 3H, CH₃), 1.25 (m, 8H, (CH₂)₄CH₃), 1.32 (m, 4H, N1CH₂CH₂(CH₂)₂), 1.90 (tt, J 7.2, 6.8 Hz, 2H, N1CH₂CH₂), 2.42 (tt, ${}^{3}J_{H-F}$ 18.8, ${}^{3}J_{H-H}$ 6.8 Hz, 2H, OCH₂CH₂CF₂), 3.82 (t, J 6.8 Hz, 2H, OCH₂CH₂CF₂), 4.34 (t, J 7.2 Hz, 2H, N1CH₂CH₂), 4.67 (s, 2H, 4-CH₂O), 7.51 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ: 13.9 (CH₃), 22.5 (CH₂CH₃), 26.4 (CH₂), 28.9 (CH₂), 29.0 (CH₂), 29.2 (CH₂), 30.2 (CH₂), 31.4 (t, ²J_{C-F} 20.9 Hz, OCH₂CH₂CF₂), 31.7 (N1CH₂CH₂), 50.4 (N1CH₂CH₂), 62.1 (t, J 4.3 Hz, OCH₂CH₂CF₂), 64.5 (4-CH₂O), 122.2 (C5-H), 144.4 (C4). MS (ESI) *m*/*z*: 594.40 ([M+Na]⁺). Anal. Calcd. for C₂₀H₂₆F₁₃N₃O requires: C, 42.04; H, 4.59; N, 7.35%. Found: C, 42.28; H, 4.65; N, 7.34%.

3.6.1.5. 1-n-Benzyl-4-(2-perfluorohexylethoxy)methyl-1H-1,2,3triazole 11. Benzyl azide (0.27 g, 2.03 mmol) to give 1-benzyl-4-(2perfluorohexylethoxy)methyl)-1H-1,2,3-triazole **11** as white flakes (0.47 g, 88%) mp 53–55 °C. IR (KBr): 3566, 3128, 3090, 2909, 2362, 1717, 1559, 1458, 1367, 1235, 1209, 1189, 1142, 1091, 1060, 1012, 913, 860, 722, 699, 653 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 2.40 (tt, ${}^{3}J_{H-F}$ 18.5, ${}^{3}J_{H-H}$ 6.8 Hz, 2H, OCH₂CH₂CF₂), 3.80 (t, *J* 6.8 Hz, 2H, OCH₂CH₂CF₂), 4.64 (s, 2H, 4-CH₂O), 5.53 (s, 2H, N1CH₂C₆H₅), 7.28 (m, 2H, N1CH₂C₆H₅-*ortho*), 7.36 (m, 1H, N1CH₂C₆H₅-*para*), 7.37 (m, 2H, N1CH₂C₆H₅-*meta*), 7.44 (s, 1H, H5). 13 C NMR (75 MHz, CDCl₃) δ : 31.4 (t, ${}^{2}J_{C-F}$ 20.9 Hz, OCH₂CH₂CF₂), 54.2 (N1CH₂C₆H₅), 62.2 (t, *J* 4.3 Hz, OCH₂CH₂CF₂), 64.5 (4-CH₂O), 122.3 (C5-H), 128.0 (N1CH₂C₆H₅-*ortho*), 128.7 (N1CH₂C₆H₅-*para*), 129.0 (N1CH₂C₆H₅-*meta*), 134.3 (N1CH₂C₆H₅-*ipso*), 144.9 (C4). MS (ESI) *m/z*: 558.33 ([M+Na]⁺, 100%). Anal. Calcd. for C₁₈H₁₄F₁₃N₃O requires: C, 40.39; H, 2.64; N, 7.85%. Found: C, 40.21; H, 2.68; N, 7.76%.

3.6.2. Preparations involving 3-(perfluoroheptylmethoxy)propyne 28 Propargyl ether **28** (0.44 g, 1.00 mmol) was made to react with the following sections.

3.6.2.1. 1-*n*-Butyl-4-(*perfluoroheptylmethoxy*)*methyl*-1*H*-1,2,3-*triazole* **12**. *n*-Butyl azide (0.20 g, 2.02 mmol) to give 1-*n*-*butyl*-4-(*perfluoroheptylmethoxy*)*methyl*-1*H*-1,2,3-*triazole* **12** as pale yellow oil (0.45 g, 84%). IR (neat): 3850, 3685, 3583, 3138, 2965, 2940, 2880, 1642, 1552, 1468, 1366, 1327, 1241, 1210, 1150, 1051, 1030, 992, 953, 808, 711, 698, 658 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 0.96 (t, *J* 7.5 Hz, 3H, CH₃), 1.36 (qt, *J* 7.5, 7.2 Hz, 2H, CH₂CH₃), 1.89 (tt, *J* 7.3, 7.2 Hz, 2H, N1CH₂CH₂), 4.03 (t, ³*J*_{H-F} 13.9 Hz, 2H, OCH₂CF₂), 4.37 (t, *J* 7.3 Hz, 2H, N1CH₂CH₂), 4.80 (s, 2H, 4-CH₂O), 7.56 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ : 13.1 (CH₃), 19.5 (CH₂CH₃), 32.1 (N1CH₂CH₂), 50.0 (N1CH₂CH₂), 65.6 (4-CH₂O), 66.7 (t, ²*J*_{C-F} 25.3 Hz, OCH₂CF₂), 122.5 (C5-H), 143.4 (C4). MS (ESI) *m/z*: 560.45 ([M+Na]⁺, 100%). Anal. Calcd. for C₁₅H₁₄F₁₅N₃O requires: C, 33.53; H, 2.63; N, 7.82%. Found: C, 33.82; H, 2.66; N, 7.53%.

3.6.2.2. 1-*n*-Hexyl-4-(*perfluoroheptylmethoxy*)*methyl*-1H-1,2,3-*triazole* **13**. *n*-Hexyl azide (0.25 g, 1.97 mmol) to give 1-*n*-hexyl-4-(*perfluoroheptylmethoxy*)*methyl*-1H-1,2,3-*triazole* **13** as a pale yellow wax (0.50 g, 88%) mp 29–31 °C. IR (neat): 3616, 3132, 2958, 2932, 2862, 2360, 1563, 1468, 1369, 1329, 1207, 1147, 1058, 1028, 992, 956, 808, 701, 659 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 0.88 (t, *J* 6.8 Hz, 3H, CH₃), 1.20–1.40 (m, 6H, (CH₂)₃CH₃), 1.91 (tt, *J* 7.2, 7.2 Hz, 2H, N1CH₂CH₂), 4.02 (t, ³J_{H-F} 13.6 Hz, 2H, OCH₂CF₂), 4.36 (t, *J* 7.2 Hz, 2H, N1CH₂CH₂), 4.80 (s, 2H, 4-CH₂O), 7.55 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ : 13.6 (CH₃), 22.2 (CH₂CH₃), 26.0 (CH₂), 30.1 (CH₂), 30.9 (N1CH₂CH₂), 50.4 (N1CH₂CH₂), 65.6 (4-CH₂O), 66.7 (t, ²J_{C-F} 25.3 Hz, OCH₂CF₂), 122.4 (C5-H), 143.5 (C4). MS (ESI) *m/z*: 589.52 ([M+H+Na]⁺, 4%), 588.58 ([M+Na]⁺, 100). Anal. Calcd. for C₁₇H₁₈F₁₅N₃O requires: C, 36.12; H, 3.21; N, 7.43%. Found: C, 36.35; H, 3.36; N, 7.37%.

3.6.2.3. 1-n-Octyl-4-(perfluoroheptylmethoxy)methyl-1H-1,2,3-triazole 14. n-Octyl azide (0.31 g, 2.00 mmol) to give 1-n-octyl-4-(perfluoroheptylmethoxy)methyl-1H-1,2,3-triazole 14 as white powder (0.49 g, 82%) mp 39–40 °C. IR (neat): 3648, 3093, 2956, 2926, 2852, 2360, 1716, 1684, 1558, 1466, 1371, 1329, 1204, 1146, 1058, 1027, 991, 957, 886, 807, 703, 660 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ: 0.87 (t, J 6.8 Hz, 3H, CH₃), 1.26 (m, 6H, (CH₂)₃CH₃), 1.32 (m, 4H, N1CH₂CH₂(CH₂)₂), 1.91 (tt, J 7.2, 6.8 Hz, 2H, N1CH₂CH₂), 4.03 (t, ³*I*_{H-F} 13.6 Hz, 2H, OCH₂CF₂), 4.36 (t, *J* 7.2 Hz, 2H, N1CH₂CH₂), 4.81 (s, 2H, 4-CH₂O), 7.55 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ: 13.9 (CH₃), 22.5 (CH₂CH₃), 26.3 (CH₂), 28.8 (CH₂), 29.0 (CH₂), 30.1 (CH₂), 31.7 (N1CH₂CH₂), 50.4 (N1CH₂CH₂), 65.6 (4-CH₂O), 66.7 (t, ${}^{2}J_{C-F}$ 25.3 Hz, OCH₂CF₂), 122.6 (C5-H), 143.5 (C4). HR-MS (ESI): Calcd. for $2 \times C_{19}H_{22}F_{15}N_3O-2H+Na$: m/z 1207.2790. Found: m/z1207.2738 (2 × M-2H+Na]+, 9%); Calcd. for $C_{19}H_{22}F_{15}N_3O$ +Na: *m*/*z* 616.1422. Found: *m*/*z* 616.1422 ([M+Na]⁺, 100).

3.6.2.4. 1-n-Nonyl-4-(perfluoroheptylmethoxy)methyl-1H-1,2,3triazole **15**. n-Nonyl azide (0.58 g, 3.43 mmol) to give 1-n-nonyl-4-(perfluoroheptylmethoxy)methyl-1H-1,2,3-triazole **15** as a white

powder (0.49 g, 81%) mp 55–56 °C. IR (neat): 3134, 2954, 2921, 2849, 1560, 1467, 1371, 1329, 1206, 1147, 1132, 1105, 1026, 990, 886, 808, 703, 661 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 0.87 (t, *J* 6.8 Hz, 3H, CH₃), 1.26 (m, 8H, (CH₂)₄CH₃), 1.32 (m, 4H, N1CH₂CH₂(CH₂)₂), 1.91 (tt, *J* 7.2, 6.8 Hz, 2H, N1CH₂CH₂), 4.03 (t, ³J_{H-F} 13.6 Hz, 2H, OCH₂CF₂), 4.36 (t, *J* 7.2 Hz, 2H, N1CH₂CH₂), 4.80 (s, 2H, 4-CH₂O), 7.55 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ : 13.9 (CH₃), 22.5 (CH₂CH₃), 26.3 (CH₂), 28.8 (CH₂), 29.0 (CH₂), 29.2 (CH₂), 30.1 (CH₂), 31.7 (N1CH₂CH₂), 50.4 (N1CH₂CH₂), 65.6 (4-CH₂O), 66.7 (t, ²J_{C-F} 25.3 Hz, OCH₂CF₂), 122.6 (C5-H), 143.5 (C4). MS (ESI) *m/z*: 1237.3260), 630.1592 ([M+Na]⁺, 57; C₂₀H₂₄F₁₅N₃O+Na requires *m/z* 630.1579). Anal. Calcd. for C₂₀H₂₄F₁₅N₃O requires: C, 39.55; H, 3.98; N, 6.92%. Found: C, 39.56; H, 3.94; N, 6.75%.

3.6.2.5. 1-n-Benzyl-4-(perfluoroheptylmethoxy)methyl-1H-1,2,3-

triazole 16. Benzyl azide (0.27 g, 2.03 mmol) to give 1-benzyl-4-(perfluoroheptylmethoxy)methyl-1H-1,2,3-triazole 16 as white needles (0.42 g, 74%) mp 70-71 °C. IR (KBr): 3648, 3126, 3089, 1716, 1684, 1647, 1558, 1473, 1458, 1370, 1326, 1250, 1202, 1142, 1129, 1111, 1065, 1018, 990, 867, 723, 698, 659 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ: 4.02 (t, ³*J*_{H-F} 13.9 Hz, 2H, OCH₂CF₂), 4.78 (s, 2H, 4-CH₂O), 5.54 (s, 2H, N1CH₂C₆H₅), 7.28 (m, 5H, N1CH₂C₆H₅-ortho), 7.37 (m, 5H, N1CH₂C₆H₅-para), 7.44 (m, 5H, N1CH₂C₆H₅-meta), 7.50 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ: 54.2 (N1CH₂C₆H₅), 65.6 (4-CH₂O), 66.8 (t, ²J_{C-F} 26.0 Hz, OCH₂CF₂), 122.7 (C5-H), 128.0 (N1CH₂C₆H₅ortho), 128.8 (N1CH₂C₆H₅-para), 129.1 (N1CH₂C₆H₅-meta), 134.2 (N1CH₂C₆H₅-*ipso*), 144.1 (C4). HR-MS (ESI) Calcd. for $2 \times C_{18}H_{12}F_{15}N_3O$ +Na: m/z 1165.1380. Found: m/z 1165.1246 $([2M+Na]^+, 100\%)$; Calcd. for C₁₈H₁₂F₁₅N₃O+Na: m/z 594.0639. Found: *m*/*z* 594.0632 ([M+Na]⁺, 40); Anal. Calcd. for C₁₈H₁₂F₁₅N₃O requires: C, 37.84; H, 2.12; N, 7.36%. Found: C, 37.91; H, 2.32; N, 7.19%.

3.7. Preparation of bis-triazolyl ethers 33-36

3.7.1. 4-Hydroxymethyl-1-n-octyl-1H-1,2,3-triazole 30

Following a modification of the method of Rostovtsev et al. [31(a)], CuSO₄·5H₂O (0.13 g, 2 mmol) and Na ascorbate (0.20 g, 1 mmol) were added to (CH₃)₂SO (20 mL) and the mixture stirred for 5 min at r.t. Propargyl alcohol (1.24 mL, 1.20 g, 22 mmol) was added dropwise with continued stirring, followed after 2 min by dropwise addition of a solution of azide 24 (3.10 g, 20 mmol) in (CH₃)₂SO (80 mL). The reaction mixture was heated in an oil bath at 65 °C for 24 h then quenched with H₂O (50 mL) and extracted with Et_2O (3 \times 50 mL). The organic phases were combined and washed sequentially with brine (2 \times 50 mL), H₂O (50 mL), and more brine (50 mL), then dried over Na₂SO₄, and evaporated to dryness under reduced pressure. The resulting brown liquid crystallized at r.t. and was flash chromatographed on silica gel using an Et₂O/light petroleum gradient (10:90 to 20:80). The major product (2.70 g) was recrystallized from pentane to give 4-hydroxymethyl-1-noctyl-1H-1,2,3-triazole **30** as white plates (2.12 g, 50%) mp 40-42 °C (lit. [35]: mp 46–49 °C). IR (Nujol) v_{max} 3310, 3208, 3132, 1336, 1297, 1257, 1220, 1212, 1147, 1055, 1043, 1017, 843, 784, 726, 675, 657 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ: 0.86 (t, J 6.5 Hz, 3H, CH₂CH₃), 1.24 (m, 6H, (CH₂)₃CH₃), 1.30 (m, 4H, N1CH₂CH₂(CH₂)₂), 1.88 (tt, J 7.2, 6.9 Hz, 2H, N1CH₂CH₂CH₂), 4.02 (s, 1H, OH), 4.33 (t, J 7.2 Hz, 2H, N1CH₂CH₂CH₂), 4.77 (s, 2H, 4-CH₂OH), 7.57 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ: 14.2 (CH₃), 22.7 (CH₂CH₃), 26.6 (CH₂), 29.05 (CH₂), 29.14 (CH₂), 30.3 (CH₂), 31.8 (N1CH₂CH₂), 50.8 (N1CH₂CH₂), 56.2 (4-CH₂OH), 122.0 (C5-H), 147.6 (C4). ¹³C NMR (75 MHz, (CD₃)₂SO) δ: 14.0 (CH₃), 22.2 (CH₂CH₃), 25.9 (CH₂), 28.5 (CH₂), 28.6 (CH₂), 29.9 (CH₂), 31.3 (N1CH₂CH₂), 49.3 (N1CH₂CH₂), 55.2 (4-CH₂OH), 122.7 (C5-H), 148.0 (C4). MS (ESI) m/z: 235.15 ([M+H+Na]⁺, 12%); 234.20 ([M+Na]⁺, 100); 212.16 ([M+H]⁺, 7). Anal. Calcd. for C₁₁H₂₁N₃O requires: C, 62.51; H, 10.04; N, 19.89%. Found: C, 62.89; H, 10.30; N, 19.98%.

3.7.2. 1-n-Octyl-4-((prop-2-ynyloxy)methyl)-1H-1,2,3-triazole 31

Following a method reported by Jones et al. [36(a)], triazolyl alcohol 30 (0.35 g, 1.7 mmol) was dissolved in dry DMF (10 mL) and crushed NaOH pellets (0.25 g, 6.3 mmol) were added. The mixture was stirred vigorously for 15 min in an ice bath under Ar, then propargyl bromide (0.20 mL of 80% solution in toluene. 0.214 g, 1.8 mmol) was added dropwise and the heterogeneous reaction mixture was stirred vigorously for 48 h, slowly warming to r.t. H₂O (30 mL) was added and the product was extracted with EtOAc (4×50 mL). The organic phases were combined and washed sequentially with 3% HCl (2×30 mL) and H₂O (30 mL). The organic phase was dried over Na₂SO₄ and evaporated to dryness under reduced pressure. The resulting dark orange liquid (0.31 g) was flash chromatographed over silica gel (Et₂O/hexane, 50:50) to generate (1-n-octyl-4-((prop-2-ynyloxy)methyl)-1H-1,2,3-triazole **31** as a pale orange oil (0.27 g, 67%), which decomposed upon attempted vacuum distillation. IR (neat) v_{max} : 3290, 3138, 2926, 2856, 2114, 1466, 1338, 1359, 1263, 1222, 1142, 1084, 1051, 1024, 942, 908, 891, 820, 784, 723, 665, 570 $\rm cm^{-1}.~^1H~NMR$ (300 MHz, CDCl₃) *δ*: 0.84 (t, *J* 6.4 Hz, 3H, CH₃), 1.27 (m, 10H, (CH₂)₅CH₃), 1.87 (m, 2H, N1CH₂CH₂), 2.45 (t, J 2.4 Hz, 1H, OCH₂CCH), 4.20 (d, J 2.4 Hz, 2H, OCH₂CCH), 4.32 (t, J 7.2 Hz, 2H, N1CH₂CH₂), 4.71 (s, 2H, 4-CH₂O), 7.54 (s, 1H, H5). ¹³C NMR (75 MHz) δ: 14.0 (CH₃), 22.5 (CH₂CH₃), 26.4 (CH₃), 28.9 (CH₂), 29.0 (CH₂), 30.2 (CH₂), 31.6 (N1CH₂CH₂), 50.4 (N1CH₂CH₂), 57.4 (4-CH₂O), 63.0 (OCH₂CCH), 74.9 (OCH₂CCH), 79.3 (OCH₂CCH), 122.6 (C5-H), 144.2 (C4). MS (ESI) *m*/*z*: 521.22 ([2M+Na]⁺, 15%), 273.03 ([M+H+Na]⁺, 17), 272.12 ([M+Na]⁺, 100), 250.27 ([M+H]⁺, 15). Anal. Calcd. for C₁₄H₂₃N₃O requires: C, 67.43; H, 9.30; N, 16.86%. Found: C, 67.46; H, 9.61; N, 17.00%.

3.7.3. General method for the reaction of propargyl ether **31** with perfluoroalkylethyl azides **17**, **18** and **19** and perfluoroheptylmethyl azide **32**

Propargyl ether **31** (0.18 g, 0.72 mmol) and perfluorobutylethyl azide **17** [13] (0.20 g, 0.69 mmol), perfluorohexylethyl azide **18** (0.28 g, 0.72 mmol), perfluorooctylethyl azide **19** [13] (0.34 g, 0.70 mmol) or perfluoroheptylmethyl azide **32** (0.31 g, 0.73 mmol) were added to a stirred suspension of sodium ascorbate (0.022 g, 0.09 mmol) and CuSO₄·5H₂O (0.012 g, 0.045 mmol) in (CH₃)₂SO (8 mL). The mixture was stirred with heating in a bath at 65 °C for 48 h. The reaction mixture was diluted with H₂O (20 mL) and the solution was extracted with CH₂Cl₂ (3 × 20 mL). The combined organic layers were washed with H₂O (2 × 10 mL), dried over Na₂SO₄, and evaporated to dryness under reduced pressure. The residue was rapidly column chromatographed on silica gel and the major fraction recrystallized.

3.7.3.1. 4-(((1-n-Octyl-1H-1,2,3-triazol-4-yl)methoxy)methyl)-1-(2-(perfluorobutyl)ethyl)-1H-1,2,3-triazole **33**. Perfluorobutylethyl azide 17 afforded the crude product as a low melting yellow solid. The solid was chromatographed on silica gel and the major fraction (0.21 g), eluted with 33:67 EtOAc/Et₂O and recrystallized from pentane, yielded 4-(((1-n-octyl-1H-1,2,3-triazol-4-yl)methox*y*)*methyl*)-1-(2-(*perfluorobutyl*)*ethyl*)-1H-1,2,3-*triazole* **33** as a white crystalline powder (0.26 g, 70%) mp 68–70 °C. IR (Nujol): *v*_{max} 2724, 2359, 1306, 1228, 1214, 1171, 1131, 1097, 1051, 1018, 989, 859, 815, 746, 724, 708 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ : 0.85 (t, J 6.6 Hz, 3H, CH₃CH₂), 1.24 (m, 6H, (CH₂)₃CH₃), 1.29 (m, 4H, N1'CH₂CH₂(CH₂)₂), 1.88 (tt, J 7.2, 6.9 Hz, 2H, N1'CH₂CH₂CH₂), 2.88 (tt, ³*J*_{H-F} 18.0, ³*J*_{H-H} 7.4 Hz, 2H, N1CH₂CH₂CF₂), 4.32 (t, *J* 7.2 Hz, 2H, N1'CH₂CH₂CH₂), 4.66 (dd, J 7.5, 7.4 Hz, 2H, N1CH₂CH₂CF₂), 4.69 (s, 2H, 4'-CH₂O), 4.70 (s, 2H, 4-CH₂O), 7.57 (s, 1H, H5' from N-alkyl ring), 7.68 (s, 1H, H5 from N-perfluorobutylethyl ring). ¹³C NMR (100.1 MHz, CDCl₃) δ : 14.1 (CH₃), 22.7 (CH₂CH₃), 26.6 (CH₂CH₂CH₃), 29.0 (CH₂), 29.1 (CH₂), 30.3 (CH₂), 31.78 (N1′CH₂CH₂CH₂), 31.82 (t, ²*J*_{C-F} 21.8 Hz, N1CH₂CH₂CF₂), 42.4 (t, ³*J*_{C-F} 5 Hz, N1CH₂CH₂CF₂), 50.7 (N1′CH₂CH₂CH₂), 63.48 (4′-CH₂O),^a 63.53 (4-CH₂O),^a 123.0 (br, C5′-H), 123.6 (C5-H) 144.3 (br, C4′), 145.1 (C4). MS (ESI) *m/z*: 561.11 ([M+Na]⁺, 9%), 540.07 ([M+2H]⁺, 26), 539.15 ([M+H]⁺, 100), 538.53 ([M]⁺, 94). [^aNote: assignments of ¹³C NMR signals at δ 63.48 and 63.53 are likely, but might be interchanged and could not be confirmed spectroscopically.] Anal. Calcd. for C₂₀H₂₇F₉N₆O requires C, 44.61; H, 5.05; N, 15.61%. Found: C, 44.79; H, 4.99; N, 15.64%.

3.7.3.2. 4-(((1-n-Octyl-1H-1,2,3-triazol-4-yl)methoxy)methyl)-1-(2-(perfluorohexyl)ethyl)-1H-1,2,3-triazole 34. Perfluorohexylethyl azide **18** afforded the crude product as a pale yellow solid, which when eluted from the silica gel column with 33:67 EtOAc/Et₂O afforded the major fraction that crystallized from pentane to yield 4-(((1-n-octyl-1H-1,2,3-triazol-4-yl)methoxy)methyl)-1-(2-(perfluorohexyl)ethyl)-1H-1,2,3-triazole 34 as white crystalline powder (0.21 g, 46%) mp 88–90 °C. IR (Nujol): v_{max} 3132, 3086, 1398, 1324, 1236, 1140, 1123, 1098, 1081, 1054, 1031, 1003, 989, 918, 850, 815, 728, 714, 703, 660, 635, 567 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ 0.85 (t, J 6.6 Hz, 3H, CH₃CH₂), 1.22 (m, 6H, (CH₂)₃CH₃), 1.28 (m, 4H, N1'CH₂CH₂(CH₂)₂), 1.86 (tt, J 7.3, 6.9 Hz, 2H, N1'CH₂CH₂CH₂), 2.79 (tt, ³*J*_{H-F} 18.0, ³*J*_{H-H} 7.5 Hz, 2H, N1CH₂CH₂CF₂), 4.31 (t, *J* 7.3 Hz, 2H, N1'CH₂CH₂CH₂), 4.65 (dd, J ca. 7.5, ca. 7.4 Hz, 2H, N1CH₂CH₂CF₂), 4.67 (s, 2H, 4'-CH₂O), 4.68 (s, 2H, 4-CH₂O), 7.58 (s, 1H, H5' from Nalkyl ring), 7.70 (s, 1H, H5 from N-perfluorohexyl ring). ¹³C NMR (100.1 MHz, CDCl₃): δ 14.0 (CH₃), 22.6 (CH₂CH₃), 26.5 (CH₂CH₂CH₃), 29.0 (CH₂), 29.1 (CH₂), 30.3 (CH₂), 31.7 (N1'CH₂CH₂CH₂), 31.9 (t, ²/_C-_F 21.7 Hz, N1CH₂CH₂CF₂), 42.4 (t, ³J_{C-F} 5.1 Hz, N1CH₂CH₂CF₂), 50.6 (N1'CH₂CH₂CH₂), 63.46 (4-CH₂O), 63.50 (4'-CH₂O), 122.9 (C5'-H), 123.6 (C5-H), 144.3 (C4'), 145.1 (C4). MS (ESI) m/z: 561.11 ([M+Na]⁺, 10%), 540.07 ([M+2H]⁺, 25), 539.11 ([M+H]⁺, 100), 538.86 ([M]⁺, 95). Anal. Calcd. for C₂₂H₂₇F₁₃N₆O requires C, 41.38; H, 4.26; N, 13.16%. Found: C, 41.58; H, 4.01; N, 13.04%.

3.7.3.3. 4-(((1-n-Octyl-1H-1,2,3-triazol-4-yl)methoxy)methyl)-1-(2-(perfluorooctyl)ethyl)-1H-1,2,3-triazole 35. Perfluorooctylethyl azide 19 afforded the crude product as a pale yellow solid, which when eluted from the silica gel column with 33:67 EtOAc/Et₂O yielded 4-(((1-n-octyl-1H-1,2,3-triazol-4-yl)methoxy)methyl)-1-(2-(perfluorooctyl)ethyl)-1H-1,2,3-triazole 35 as white crystalline powder (0.21 g, 41%) mp 104–106 °C. IR (Nujol): v_{max} 3131, 3083, 2426, 2252, 1905, 1694, 1562, 1418, 1375, 1336, 1200, 1146, 1099, 1056, 1003, 989, 958, 907, 852, 824, 774, 736, 690, 654, 560 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ : 0.85 (t, J 6.8 Hz, 3H, CH₃CH₂), 1.25 (m, 6H, (CH₂)₃CH₃), 1.30 (m, 4H, N1'CH₂CH₂(CH₂)₂), 1.88 (tt, J 7.2, 6.4 Hz, 2H, N1′CH₂CH₂CH₂), 2.81 (tt, ³J_{H-F} 17.9, ³J_{H-H} 7.5 Hz, 2H, N1CH₂CH₂CF₂), 4.33 (t, J 7.2 Hz, 2H, N1'CH₂CH₂CH₂), 4.66 (dd, J 7.7, 7.4 Hz, 2H, N1CH₂CH₂CF₂), 4.70 (s, 4H, 4-CH₂O and 4'-CH₂O), 7.58 (s, 1H, H5' from N-alkyl ring), 7.69 (s, 1H, H5 from Nperfluorooctyl ring). ¹³C NMR (100.1 MHz, CDCl₃) δ : 14.1 (CH₃), 22.7 (CH₂CH₃), 26.6 (CH₂CH₂CH₃), 29.0 (CH₂), 29.1 (CH₂), 30.4 (CH₂), 31.8 (N1′CH₂CH₂CH₂), 31.9 (t, ²*J*_{C-F} 22.0 Hz, N1CH₂CH₂CF₂), 42.4 (t, ³J_{C-F} 4.1 Hz, N1CH₂CH₂CF₂), 50.7 (N1′CH₂CH₂CH₂), 63.51 (4-CH₂O) 63.54 (4'-CH₂O), 123.0 (C5'-H), 123.7 (C5-H) 144.4 (C4'), 145.2 (C4). MS (ESI) m/z: 561.11 ([M+Na]⁺, 10%), 540.07 ([M+2H]⁺, 25), 539.11 ([M+H]⁺, 100), 538.86 (M⁺, 95). Anal. Calcd. for C₂₄H₂₇F₁₇N₆O₂ requires C, 39.03; H, 3.69; N, 11.38%. Found: C, 38.88; H, 3.60; N, 11.20%.

3.7.3.4. 4-(((1-n-Octyl-1H-1,2,3-triazol-4-yl)methoxy)methyl)-1-((perfluoroheptyl)methyl)-1H-1,2,3-triazole 36. Perfluoroheptylmethyl azide 32 afforded the crude product as pale yellow solid, which when eluted from the silica gel column with 33:67 EtOAc/ Et₂O yielded 4-(((1-n-octyl-1H-1,2,3-triazol-4-yl)methoxy)methyl)-1-(2-(perfluoroheptyl)methyl)-1H-1,2,3-triazole 36 as white crystalline powder (0.15 g, 31%) mp 114–117 °C. ¹H NMR (400 MHz, CDCl₃) δ: 0.86 (t, J 6.5 Hz, 3H, CH₃CH₂), 1.25 (m, 6H, (CH₂)₃CH₃), 1.30 (m, 4H, N1'CH₂CH₂(CH₂)₂), 1.89 (tt, J 6.7, 6.7 Hz, 2H, N1'CH₂CH₂CH₂), 4.33 (t, J 7.2 Hz, 2H, N1'CH₂CH₂CH₂), 4.71 (s, 2H, 4'-CH₂O), 4.73 (s, 2H, 4-CH₂O), 5.05 (t, ³J_{H-F} 14.8 Hz, NCH₂CF₂), 7.58 (br s, 1H, H5' from N-alkyl ring), 7.78 (s, 1H, H5 from Nperfluoroheptylmethyl ring). ¹³C NMR (100.1 MHz, CDCl₃) δ : 14.1 (CH₃), 22.7 (CH₂CH₃), 26.6 (CH₂CH₂CH₃), 29.0 (CH₂), 29.1 (CH₂), 30.3 (CH₂), 31.8 (N1'CH₂CH₂CH₂), 49.4 (t, ²J_{C-F} 23 Hz, NCH₂CF₂), 50.8 (N1'CH₂CH₂CH₂), 63.5 (2C, 4-CH₂O and 4'-CH₂O), 123.1 (br, C5'-H), 124.7 (C5-H) 144.4 (br, C4'), 145.8 (C4). HR-MS (ESI) (Found *m*/*z*: 1371.3598 ([2M+Na]⁺, 100%), 697.1727 ([M+Na]⁺, 28). $C_{22}H_{27}N_6OF_{13}$ requires m/z: 1371.3598 ([2M+Na]⁺), 697.1748 ([M+Na]⁺). Anal. Calcd. for C₂₂H₂₅F₁₅N₆O₂ requires C, 39.18; H, 3.74; N, 12.46%. Found: C, 39.18; H, 3.58; N, 12.25%.

3.8. Preparation of 4-(((1-n-nonyl-1H-1,2,3-triazol-4yl)methoxy)methyl)-1-(2-(perfluorooctyl)ethyl)-1H-1,2,3-triazole **38**

3.8.1. 4-Hydroxymethyl-1-n-nonyl-1H-1,2,3-triazole 37

Propargyl alcohol 29 (0.58 mL, 10 mmol) was added dropwise to a solution of sodium ascorbate (0.178 g, 0.90 mmol) and CuSO₄·5H₂O (0.112 g, 0.45 mmol) in (CH₃)₂SO (25 mL). The bright yellow-green solution was stirred for 1 min before nonyl azide 25 (1.70 g, 10 mmol) was added dropwise, then stirring was continued in a bath at 70 °C for 36 h. The reaction mixture was diluted with water (75 mL), extracted with EtOAc (3×50 mL), the combined organic layers were washed with brine $(2 \times 25 \text{ mL})$, dried over Na₂SO₄, and evaporated to dryness under reduced pressure. The residue was flash chromatographed (silica gel, $Et_2O/$ EtOAc 1:1) to give 4-hydroxymethyl-1-n-nonyl-1H-1,2,3-triazole 37 [38] (1.87 g, 79%) mp 55–57 °C (Et₂O/light petroleum). IR (Nujol): 3301, 3195, 3131, 2922, 2853, 1463, 1376, 1218, 1144, 1055, 1046, 1018, 724 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ : 0.84 (t, J 6.6 Hz, 3H, CH₂CH₃), 1.22 (m, 8H, (CH₂)₄CH₃), 1.27 (m, 4H, N1CH₂CH₂(CH₂)₂), 1.85 (tt, J 7.2, 6.9 Hz, 2H, N1CH₂CH₂CH₂), 4.25 (s, 1H, OH), 4.29 (t, J 7.2 Hz, 2H, N1CH₂CH₂CH₂), 4.74 (s, 2H, 4-CH₂OH), 7.55 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ: 14.0 (CH₃), 22.5 (CH₂CH₃), 26.4 (CH₂), 28.9 (CH₂), 29.1 (CH₂), 29.2 (CH₂), 30.1 (CH₂), 31.7 (N1CH₂CH₂), 50.4 (N1CH₂CH₂), 55.9 (4-CH₂OH), 121.7 (C5-H), 147.7 (C4). MS (ESI) m/z: 473.11 ([2M+Na]⁺, 22%), 263.99 ([M+K]⁺, 7), 249.22 ([M+H+Na]⁺, 16), 248.14 ([M+Na]⁺, 100), 226.18 ([M+H]⁺, 13). Anal. Calcd. for C₁₂H₂₃N₃O requires C, 63.99; H, 10.29; N, 18.65%. Found: C, 63.59, H, 10.37; N, 18.29%.

3.8.2. 1-n-Nonyl-4-((prop-2-ynyloxy)methyl)-1H-1,2,3-triazole 38

Alcohol 37 (0.53 g, 2.35 mmol) was dissolved in DMF (10 mL) and powdered NaOH pellets (0.32 g, 8.0 mmol) were added. The contents were stirred in a salt-ice bath for 10 min then propargyl bromide (0.25 mL, 2.82 mmol) was added dropwise. The reaction mixture allowed to stir for 24 h, gradually warming to r.t. The reaction mixture was partitioned between Et₂O (30 mL) and water (50 mL) and the aqueous layer extracted with more Et₂O $(3 \times 30 \text{ mL})$. The combined organic extracts were dried over Na₂SO₄, and evaporated to dryness under reduced pressure. The residue was flash chromatographed (silica gel, Et₂O/light petroleum 1:1) to give 1-n-nonyl-4-((prop-2-ynyloxy)methyl)-1H-1,2,3triazole 38 as needles (0.42 g, 68%) mp 77-79 °C (Et₂O/light petroleum). IR (Nujol): 3278, 3089, 2922, 2853, 2108, 1460, 1376, 1224, 1153, 1103, 1059, 1027, 935, 730 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ: 0.86 (t, J 6.9 Hz, 3H, CH₂CH₃), 1.24 (m, 8H, (CH₂)₄), 1.30 (m, 4H, (CH₂)₂), 1.89 (tt, J 7.2, 7.0 Hz, 2H, CH₂CH₂N), 2.46 (t, J 2.4 Hz, 1H, OCH₂CCH), 4.22 (d, J 2.4 Hz, 2H, OCH₂CCH), 4.34 (t, J 7.2 Hz, 2H, CH₂N), 4.74 (d, J 0.5 Hz, 2H, 4-CH₂O), 7.56 (s, 1H, H5). ¹³C NMR (75 MHz, CDCl₃) δ: 14.0 (CH₃), 22.6 (CH₂CH₃), 26.4 (CH₃), 28.9 (CH₂), 29.1 (CH₂), 29.3 (CH₂), 30.2 (CH₂), 31.7 (N1CH₂CH₂), 50.5 (N1CH₂CH₂), 57.5 (4-CH₂O), 62.9 (OCH₂CCH), 74.9 (OCH₂CCH), 79.2 (OCH₂CCH), 122.6 (C5-H), 144.1 (C4). MS (ESI) m/z: 286.08 ([M+Na]⁺, 39%), 264.24 ([M+H]⁺, 46). Anal. Calcd. for C₁₅H₂₅N₃O requires C. 68.40: H. 9.57: N. 15.95%. Found: C. 68.70: H. 9.84: N. 15.86%.

3.8.3. 4-(((1-n-Nonyl-1H-1,2,3-triazol-4-yl)methoxy)methyl)-1-(2-(perfluorooctyl)ethyl)-1H-1,2,3-triazole 39

Propargyl ether **38** (0.263 g, 1.0 mmol) and perfluorooctylethyl azide 19 (0.49 g, 2.0 mmol) were added to a stirred suspension of sodium ascorbate (0.018 g, 0.09 mmol) and CuSO₄·5H₂O (0.012 g, 0.048 mmol) in $(CH_3)_2$ SO (15 mL). The mixture was stirred with heating in a bath at 85 °C for 48 h. The reaction mixture was diluted with H₂O (30 mL), extracted with EtOAc (3×30 mL), and the combined organic layers were washed with brine $(2 \times 20 \text{ mL})$, dried over Na₂SO₄, and evaporated to dryness under reduced pressure. The residue was flash chromatographed (silica gel, Et₂O/ EtOAc 1:1) to yield 4-(((1-n-nonyl-1H-1,2,3-triazol-4-yl)methoxy)methyl)-1-(2-(perfluorooctyl)ethyl)-1H-1,2,3-triazole 39 as white prisms (0.55 g, 73%) mp 117-119 °C (pentane). IR (Nujol): 3131, 3081, 2924, 2724, 1563, 1459, 1376, 1336, 1220, 1200, 1146, 1116, 1099, 1055, 1041, 1031, 1003, 988, 956, 850, 723 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ: 0.86 (t, J 6.9 Hz, 3H, CH₃CH₂), 1.25 (m, 8H, (CH₂)₄CH₃), 1.31 (m, 4H, N1'CH₂CH₂(CH₂)₂), 1.90 (tt, J 7.3, 6.9 Hz, 2H, N1/CH₂CH₂CH₂), 2.82 (m, 2H, N1CH₂CH₂CF₂), 4.34 (t, [7.3 Hz, 2H, N1/CH₂CH₂CH₂), 4.68 (dd, J 7.5, 7.4 Hz, 2H, N1CH₂CH₂CF₂), 4.72 (br s, 4H, 4-CH₂O and 4'-CH₂O), 7.61 (br s, 1H, H5' from N-alkyl ring), 7.71 (br s, 1H, H5 from N-perfluorooctyl ring). ¹⁹F NMR (282 MHz, CDCl₃) δ: -81.2 (t, 3F), -114.5 (t, 2F), -122.2 (m, 6F), -123.1 (s, 2F), -123.8 (s, 2F), -126.4 (m, 2F). ¹³C NMR (75 MHz, CDCl₃) *δ*: 14.0 (CH₃), 22.6 (CH₂CH₃), 26.4 (CH₂CH₂CH₃), 28.9 (CH₂), 29.1 (CH₂), 29.3 (CH₂), 30.2 (CH₂), 31.75 (N1'CH₂CH₂CH₂), 31.80 (t, ${}^{2}J_{C-F}$ 22.1 Hz, N1CH₂CH₂CF₂), 42.3 (br, N1CH₂CH₂CF₂), 50.6 (N1'CH₂CH₂CH₂), 63.3 (4'-CH₂O) 63.4 (4-CH₂O), 122.9 (C5'-H), 123.5 (C5-H) 144.2 (C4'), 145.0 (C4). MS (ESI) m/z: 791.43 ([M+K]⁺, 22%), 776.06 ([M+H+Na]⁺, 23), 774.98 ([M+H]⁺, 100), 752.87 (M⁺, 16). Anal. Calcd. for C₂₅H₂₉F₁₇N₆O requires C, 39.90; H, 3.88; N, 11.17%. Found: C, 39.95; H, 3.67; N, 11.09%.

Acknowledgements

This work was financially supported by Australian Research Council Discovery Project Grant DP0346614 and the University of New South Wales. High resolution mass spectrometric analysis for this work was carried out at the Bioanalytical Mass Spectrometry Facility, UNSW, and was supported in part by infrastructure funding from the New South Wales Government as part of its coinvestment in the National Collaborative Research Infrastructure Strategy. A.I.M. was generously supported through an Australian Research Council Endeavour Postdoctoral Research Fellowship Award. Ms. Berta Litvak is thanked for her English language translation of Ref. [38].

Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.jfluchem.2011.07.002.

References

- [1] E. Kissa, Fluorinated Surfactants. Synthesis, Properties, Applications, Marcel Dekker, New York, 1994.
- [2] P.D.I. Fletcher, Specialist Surfactants: Fluorinated and Semi-Fluorinated Surfactants, Blackie, London, 1997.
- [3] T. Hiyama, Organofluorine Compounds: Chemistry and Applications, Springer-Verlag, Berlin, 2000.
- [4] M.P. Krafft, F. Giulieri, J.G. Riess, Angew. Chem., Int. Ed. Engl. 32 (1993) 741-743.
- [5] J.G. Riess, J. Drug Target 2 (1994) 455-468.
- [6] M.P. Krafft, J.G. Riess, Cell. Mol. Biol. Lett. 1 (1996) 459-468.
- [7] M.P. Krafft, J.G. Riess, Biochimie 80 (1998) 485-489.
- [8] K. Matsuoka, Y. Moroi, Curr. Opin. Coll. Interface Sci. 8 (2003) 227-235.
- [9] Y. Kondo, N. Yoshino, Curr. Opin. Coll. Interface Sci. 10 (2005) 88-93.
- [10] J.N. Slaughter, K.M. Schmidt, J.L. Byram, S. Mecozzi, Tetrahedron Lett. 48 (2007) 3879-3882.
- [11] S. Cosgun, M. Özer, F. Hamdoune, C. Gerardin, S. Thiebaut, B. Henry, J. Amos, L. Rodehüser, C. Selve, J. Fluorine Chem. 107 (2001) 375-386.
- [12] A.R. Katritzky, T.L. Davis, G.W. Rewcastle, G.O. Rubel, M.T. Pike, Langmuir 4 (1988) 732-735.
- [13] J.B. Nivet, R. Bernelin, M. Le Blanc, J.G. Riess, Eur. J. Med. Chem. 27 (1992) 891-898. [14] Y.M. Wu, J. Deng, X. Fang, Q.Y. Chen, J. Fluorine Chem. 125 (2004) 1415-1423, see
- for close analogy.
- [15] E. Mayot, C. Gérardin-Charbonnier, C. Selve, J. Fluorine Chem. 126 (2005) 715-720.
- [16] Z. Kaleta, O. Egyed, T. Soós, Org. Biomol. Chem. 3 (2005) 2228-2230.
- S.M. Vyas, J. Turánek, P. Knötigová, A. Kasná, V. Kvardová, V. Koganti, S.E. Rankin, [17]B.L. Kuntson, H.-J. Lehmler, New J. Chem. 30 (2006) 944-951.
- [18] O.M. Martin, S. Mecozzi, Tetrahedron 63 (2007) 5539-5547. [19] K. Matsuoka, T. Yoshimura, T. Shikimoto, J. Hamada, M. Yamawaki, C. Honda, K.
- Endo, Langmuir 23 (2007) 10990-10994.
- [20] W. Zhang, C. Cai, Chem. Commun. (2008) 5686-5694.
- [21] W. Zhang, D.P. Curran, Tetrahedron 62 (2006) 11837-11865 [22] (a) A. Gheorghe, E. Cuevas-Yañez, J. Horn, W. Bannwarth, B. Narsaiah, O. Reiser,
- Synlett (2006) 2767-2770:
- (b) T. Chinnusamy, O. Reiser, ChemSusChem 3 (2010) 1040-1042.
- [23] Y.-W. Zhu, W.-B. Yi, C. Cai, J. Fluorine Chem. 132 (2011) 71-74.
- [24] R.W. Read, X.-B. Wang, Chiang Mai, S.Ci. 36 (2009) 247–257.
 [25] Y. Huang, R.W. Read, X. Wang, Aust. J. Chem. 63 (2010) 802–807.
- [26] R.W. Read, X. Wang, A structure-function study of the surface tension changes of *m*-xylene in the presence of fluorous 1*H*-1,2,3-triazoles and tetrazoles, J. Fluorine Chem., doi:10.1016/j.jfluchem.2011.07.030.
- [27] M. Napoli, C. Fraccaro, A. Scipioni, J. Fluorine Chem. 51 (1991) 103-115.
- [28] C.S. Rondestvedt Jr., G.L. Thayer Jr., J. Org. Chem. 42 (1977) 2680–2683.
- [29] F. Szonyi, A. Cambon, J. Fluorine Chem. 42 (1989) 59-68.
- [30] D.J. Wardrop, J. Fritz, Org. Lett. 8 (2006) 3659-3662, supplementary information. [31] (a) V.V. Rostovtsev, L.G. Green, V.V. Fokin, K.B. Sharpless, Angew. Chem., Int. Ed.
 - 41 (2002) 2596-2599: (b) C.W. Tornøe, C. Christensen, M. Meldal, J. Org. Chem. 67 (2002) 3057-3064; (c) R. Huisgen, in: A. Padwa (Ed.), 1,3-Dipolar Cycloaddition Chemistry, Wiley, New York, 1984, pp. 1-176; (d) V.O. Rodionov, V.V. Fokin, M.G. Finn, Angew. Chem., Int. Ed. 44 (2005)
- 2210-2215.
- [32] A.K. Feldman, B. Colasson, V.V. Fokin, Org. Lett. 22 (2004) 3897-3899. [33] A.V. Maksikova, E.S. Serebryakova, L.G. Tikhonova, L.I. Vereshchagin, Chem.
- Heterocycl. Compd. (1980) 1284-1285. [34]
- S. Özçubukçu, E. Ozkal, C. Jimeno, M.A. Pericàs, Org. Lett. 11 (2009) 4680-4683. [35] C.O. Kappe, E. Van der Eycken, Chem. Soc. Rev. 39 (2010) 1280-1290.
- [36] (a) G.B. Jones, J.M. Wright, G. Hynd, J.K. Wyatt, P.M. Warner, R.S. Huber, A. Li, M.W. Kilgore, R.P. Sticca, R.S. Pollenz, J. Org. Chem. 67 (2002) 5727-5732;
- (b) B. Jin, Q. Liu, G.A. Sulikowski, Tetrahedron 61 (2005) 401-408.
- [37] M. Abe, K. Morikawa, K. Ogino, H. Sawada, T. Matsumoto, M. Nakayama, Langmuir 8 (1992) 763-764.
- [38] A.V. Maksikova, G.P. Sukjanov, L.I. Vereshchagin, G.A. Gareev, Izv. Vyssh. Uchebn. Zaved., Khim. Khim. T 27 (1984) 172-177 (English translation).